On soil genesis in temperate humid climate. II. The behaviour of the non-clay fraction in some soil groups

J. VAN SCHUYLENBORGH

Department of Regional Pedology, Mineralogy and Geology, Agriculture University, Wageningen, Netherlands

Summary

The conclusions drawn in part I of this series could be confirmed in all cases but one. This exception was discussed.

The final conclusion could be drawn that, although aluminium migrates with clay (if there is clay migration), the translocation of aluminium and iron is fundamentally caused by the same process.

1. Introduction

In part I of this investigation (VAN SCHUYLENBORGH, 1962) analytical data on the composition of the profiles of several soil groups were given; also the clay separates were analyzed. The results permitted certain conclusions on the genesis of the soils. One result was that in podzols, grey-brown podzolic-, brown podzolic-, and gley soils, the iron was more mobile than aluminium. In the acid brown earths, however, the reverse was true.

Because it is imaginable that the translocation of iron is a different process than that of aluminium (this can be transported by clay migration), it is necessary to investigate the non-clay fraction of the soil. It is not probable that this fraction moves by mechanical forces except perhaps the very fine part of the silt fraction. It is the purpose of this paper to study the translocations of iron and aluminium in the non-clay part of the soil.

2. Methods

The SiO₂-, Al₂O₃-, and Fe₂O₃-contents of the non-clay fraction of the soils were calculated from the soil- and clay composition and the clay content. The results, with the silica sesquioxide ratios, are shown in the TABLE.

3. Results and discussion

The TABLE shows that the conclusions drawn in part I of this publication series and recapitulated in the introduction, are comfirmed in all cases but one. This means that the translocation of iron and aluminium is fundamentally the same process.

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Soil group	Profile	Horizon	Percentages			Molar ratios			
			SiO ₂	Al_2O_3	Fe ₂ O ₈	SiO ₂ / R ₂ O ₃	SiO ₂ / Al ₂ O ₈	SiO_2/Fe_2O_3	Al ₂ O ₃ / Fe ₂ O ₃
Podzol (Spodosol)	11	A ₁	84,9	1,64	0,41	76	88	545	6,18
		A_2	92,7	2,00	0,35	71	79	702	8,90
		B_{2h}	83,2	3,10	1,40	35	46	157	3,46
		\mathbf{B}_{2ir}	84,1	4,56	1,54	26	31	141	4,50
		C	85,7	4,64	1,19	27	31	193	6,15
	III	A ₁	93,7	0,9 6	0,17	133	148	1268	8,50
		A_2	96,4	0,86	0,10	169	191	2678	14,0
		$\mathbf{B}_{2\mathbf{h}}$	89,1	2,24	0,31	62	68	782	11,6
		B _{2ir}	90,7	3,08	0,88	42	50	275	5,49
		С	94,3	2,32	0,46	61	69	542	7,84
	IV	A ₁	87,5	0,97	0,34	126	153	694	4,50
		A_2	89,3	1,11	0,32	115	136	744	5,45
		$\mathbf{B_{2h}}$	81,9	1,95	0,66	59	71	333	4,65
		B _{2ir}	83,6	2,21	0,74	53	64	303	4,72
		С	86,3	1,20	0,37	102	122	625	5,14
Grey-brown podzolic (Alfisol: Udalf)	VI	A ₁	76,2	2,48	1,10	41	52	181	3,43
		A_2	78,7	2,98	1,22	36	45	164	3,63
		$\mathbf{B_1}$	74,9	3,00	1,05	35	42	189	4,46
		$\mathbf{B_2}$	66,3	3,32	0,97	29	34	184	5,43
		C	68,6	3,42	1,01	29	34	190	5,66
	VII	$\mathbf{A_2}$	74,5	4,24	0,87	26	30	230	7.76
		B ₁	72,7	4,70	1,37	22	26	141	5,36
		B_{21}	69,2	4,59	1,55	21	26	119	4,64
		B_{22}	68,4	4,98	1,97	19	23	89	3,81
	XV 1		84,6	1,59	1,12	62	90	202	2,23
		\mathbf{A}_2	84,9	1,48	0,83	72	98	272	2,79
		\mathbf{B}_1	75,5	2,44	1,24	40	53	161	3,06
		\mathbf{B}_2	62,7	3,99	2,72	19	27	62	2,30
Brown podzolic (Spodosol)	х	A ₁	83,6	1,92	0,85	58	74	263	3,54
		\mathbf{B}_2	86,1	2,82	0,84	44	52	276	5,32
		B	87,3	3,17	0,96	39	47	243	5,18
		C	89,7	3,54	1,19	36	43	202	4,70
Acid brown earth (Inceptisol)	VIII	A1	86,6	1,60	0,57	75	92	401	4,36
		(B) ₁	90,2	1,64	0,54	77	93	442	4,73
		(B) ₃	90,0	1,92	0,57	67	80	417	5,23
		C	91,5	2,09	0,67	62	74	363	4,87
	IX	A ₁	82,4	2,48	1.28	42	40	172	3,04
		(B) ₁	87,3	2,67	1,30	42	36	180	3,24
		(B) ₃	90,5	3,73	1,33	34	41	182	4,41
		Ċ Ĩ	88,3	3,11	0,99	40	48	237	4,92
Gley soil (Alfisol: Aqualf)	XI	A ₁	76,8	5,14	1,73	21	25	119	
		A_{2g}	79,4	5,65	2,23	19	24	95	4,67
		B_{1g}	72,0	6,82	3,60	13	18		4,00
		\mathbf{B}_{2g}^{1g}	61,2	5,87	3,24	13	18	53 51	2,97 2,85
	XIV	-2s A ₁	68,9	4,17	0,82	25			
	774 V	A_1 A_{2g}	72,3	4,17	1,13	25 25	28	230	8,20
		B_{1g}	71,7	4,60	1,13	23	29	172	5,71
		\mathbf{B}_{2g}^{1g}	63,8	5,54	1,27	23 17	27	149	5,62
		G ^{2g}	72,8	5,47	1,44		20	118	6,03
		5	14,0	J, T /	1,40	19	23	135	6,00

TABLE. Data on the non-clay fraction of various soil profiles; contents in percentages of oven-dry soil

¹ Prof. XV was not included in Part I of this paper-series; it is an intergrade between a grey-brown and red-yellow podzolic soil in river loam.

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Although aluminium migrates with the clay minerals, if there is clay eluviation, the weathering reactions resulting in the liberation of Al_2O_3 and Fe_2O_3 and the subsequent translocation by organic acids, are more important.

The one exception is profile VI, being a grey-brown podzolic soil developed in loess. The non-clay fraction shows a more or less constant Fe_2O_3 -content (see also the SiO_2/Fe_2O_3 -ratio), whereas the Al_2O_3 -content of the different horizons definitely increases with depth. The clay fraction (see part I) shows just the reverse. This is difficult to understand. Apparently, the Al-silicates of the clay fraction are stable under the conditions of formation and the Al-minerals of the non-clay fraction are not. For the iron compounds in both fractions the reverse must then be true. The mineralogical examination of the sand fraction did not give sufficient support for this in so far that the percentage opaque minerals (for the greatest part magnetite) plus limonite is 55, 74, 35, 44 and 49 of the heavy fraction for the A₁, A₂, B₁ and C respectively, which is fairly well in agreement with the Fe₂O₃-contents of the non-clay fraction. When we suppose that these minerals are stable under the formation conditions, this would account for the immobility of iron in the sandfraction.

Another possibility would be the migration of the very fine part of the silt fraction. This is, however, not very likely because the percentage silt decreases with depth, viz. 74,0; 77,3; 74,7; 70,5 and 72,2 in the A₁, A₂, B₁, B₂ and C-horizon respectively.

The profiles XI and XIV being developed in loess also, do not show such phenomena, probably because of the quite different soil forming conditions. Here the alternating reduction oxidation conditions in cooperation with the organic acids account for the translocation of iron.

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LITERATURE

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