THE EFFECT OF AIR-DRYING OF SOIL SAMPLES UPON SOME PHYSICAL SOIL PROPERTIES ¹)

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SUMMARY

Results obtained with tropical soils show that air-drying of samples gives an erroneous picture of some physical properties (mechanical composition and consistency values) of these soils. When the soils are not dried at the air before analysing they appear to be much heavier than analysed after the usual drying process. Likewise the consistency values are lowered appreciably and the plastic number (or plasticity index) changes from that of plastic clay soils to that of coarse sands (see table 3). The origin of these irreversible changes in the physical state of soils by air-drying are discussed. The author concludes that it is not allowed to dry samples of tropical soils before analysing.

INTRODUCTION

It is customary to dry soil samples at the air before analysing. This drying may lead to changes in the chemical state of the soil, as was emphasized in a recent paper (VAN SCHUYLENBORCH, 1953). As a consequence, the process of air-drying leads to an erroneous picture of the chemical properties of the soil and this is one of the reasons why, in many cases, soil analysis fails to be a reliable basis for fertilizing advices.

In this paper it will be shown that also physical characteristics, such as the mechanical composition and the ATTERBERG consistency values of special tropical soils are profoundly affected by the process of drying.

MATERIAL AND METHODS

Samples were taken from three profiles of mountain soils in West Java, Indonesia, in the neighbourhood of the village Leuwiliang. The profiles are lying at an altitude of 1000 m and are grown with evergreen mountain forest. The rainfall is excessive and amounts to 7 m a year. The mean annual temperature at this altitude is about 25 °C. The profiles are developed in andesitic tuff and are of the type of brown forest, possibly weakly podzolized, soils.

The following is a typical profile.

A_{00}	0-1	cm.	Forest litter.
\mathbf{A}_{0}	1-3	,,	Very loose, crumbly, fine mull. Org. matter content 48.8 %.
A	3–13	· ,,	Dark reddish brown (MUNSELL 5 YR $- 3/2$), very porous, nutty structure, rich in humus (35.0 %), with many roots.
	13-41	,,	Brown-dark brown (MUNSELL 10 YR $- 4/3$), fine crumbly, with many roots and rich in humus (20.3 %).
	4176	"	Deep strongly brown (MUNSELL 7.5 YR $-$ 5/6), rather compact. Org. matter content 7.2 %. Practically no roots.
B	76-100	"	Brown to dark brown (MUNSELL 7.5 YR $- 4/4$), loose, loamy; org. matter content 4.2 %.
С	> 100	"	Yellowish brown, very porous, strongly weathered tuff. Org. matter content 1.6% .

The mineralogical composition of the fraction $> 50 \ \mu$ of the various horizons of this profile is given in table 1.

1) Received for publication June 10, 1953.

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Horizon	Opaque	Hypersthene	Oligoclase	Quartz	SiO ₂ 1)	Volcanic glass	Muscovite	Gibbsite	Miscellaneous	Remarks
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	5 5 1 6 5 2	3 5 4 5 7	3 6 3 33 31	3 4 6 6 -	31 43 12 5 -	44 27 15 14 15 24	2 1 - 2 -	- - 31 15	9 9 56 60 11 21	1) This silicic acid is present as small tubes and was deposited around root hairs. X-ray analysis shows it to be amorphous just as the volcanic glass.

Table 1. Mineral counts of the fraction > 50 μ of the typical profile of a mountain soil from West-Java.

The profile description and the mineralogical analysis indicate the following characteristics. The organic matter content is very high and decreases gradually with the depth. There is an accumulation of SiO_2 in the superficial soil layers and an accumulation of gibbsite in the B-horizon, which indicates a podzolizing process. It is, however, remarkable that there is no indication of an accumulation or translocation of iron (for a detailed mineralogical description, see: VAN RUMMELEN, 1953).

Samples from these profiles were taken and stored in paraffine paper bags, sealed as tight as possible and transported to the laboratory as soon as possible. After arriving the samples were divided into four portions. The first portion was kept moist; the second portion was oxidized with H_2O_2 ; the third portion was dried at the air and the fourth portion was dried at the air and coxidized with H_2O_2 . Of all portions the mechanical composition was determined according to MOHR (1910), with ammonia as peptizing agent.

To check the validity of this agent for the dispersion of tropical soils, the effect of several electrolytes was investigated by determining the fraction $< 0.3 \mu$. Tre greater this fraction the better the peptization. The results are represented in table 2.

$< 0.5 \mu$ or some intermediate			1	1	1 😅
Soil	Na4P2O7	HOłHN	Na2C2O4	(NaPO ₃)6	Na ₄ Fe(CN)
Indonesian soils : Young brown lahar soil (Bogor) Strongly weathered brown lahar soil (Bogor) Lateritic clay soil (Kalimati) Laterite clay soil (Depok) Dacitic tuff red earth (Vorstenlanden)	29.0 17.2 20.5 73.9 37.1	26.0 15.4 16.6 74.7 30.4	24.2 12.8 20.0 12.0 33.8	12.8 1.5 5.5 19.9 0.2	2.5 1.0 0.1 0.0 0.1
Dutch soils : Sea clay soil (Groetpolder) Sea clay soil (Westerwijtwerd) Riverclay soil (Zaltbommel) Heavy loam (Markelo) Loessial soil (Voerendaal)	$ 19.0 \\ 17.8 \\ 26.8 \\ 34.8 \\ 14.5 $	11.0 16.5 17.3 25.5 12.4	$17.1 \\ 16.7 \\ 22.5 \\ 31.4 \\ 13.5$	3.1 15.2 18.0 12.8 13.5	13.2 17.0 11.1 0.0 12.0

Table 2. The effect of electrolytes on dispersion, measured by the clay content (fraction $< 0.3 \ \mu$) of some Indonesian and Dutch soils.

The table shows that sodium pyrophosphate is the best peptizing electrolyte both for tropical and for Dutch soils, followed by ammonia for Indonesian and by sodium oxalate for Dutch soils. It can be concluded that the use of ammonia for the dispersion of tropical soils is justified, because the values do not deviate much from the values obtained with sodium pyrophosphate.

The ATTERBERG consistency values of the moist and air-dried portions of the soil samples were determined according to MOHR (1915). The results, together with those of the mechanical analyses, are given in table 3.

RESULTS

Generally spoken, the table shows that the effect of air-drying is that of changing a clay soil into a sand (compare the data for moist, oxidized and air-dried + oxidized). This is in accordance with the change in plastic number (from a plastic clay into a coarse sand, rich in organic matter) and the water-holding capacity.

The change in the mechanical composition is demonstrated especially in the clay fraction $(< 2 \mu)$ and in the coarse fraction $(> 50 \mu)$, the medium sized fractions being not altered very strongly, at least when comparing the results of the moist, oxidized and air-dried + oxidized portions. When the results of the moist and air-dried portions of the samples are compared, the process of air-drying can be detected in all fractions, the finer fractions (beyond 20 μ) being diminished and the coarser fractions being enlarged. Furthermore it appears that oxidation with H₂O₂ hardly changed the effect of air-drying: the fractions $< 2 \mu$ and $> 50 \mu$ of the air-dried and oxidized portions of the samples do not differ much from those of the air-dried portion, except in one case (II, 1-5 cm). This means, that the effect of air-drying is not an alteration of the organic matter but rather a change in the inorganic matter of the soil.

DISCUSSION

The classification of soils is largely based on the mechanical composition of the soil and that of Indonesian soils also on the ATTERBERG consistency values. It is therefore necessary to have methods giving exact data of the true relationships in the soil. The results show, however, that air-drying of the soil before analysis gives a completely erroneous picture of both the mechanical composition and of the consistency values, at least for Indonesian mountain soils. It is probable that the physical properties of other soils also are affected by the process of air-drying.

It can be remarked *a priori* that the effect of air-drying will depend upon three factors: 1 the climate; 2 the inorganic composition of the soil; 3 the organic matter of the soil.

The physical properties of soils in arid climates will not be affected strongly by air-drying, because they are completely dry already for long periods during the year. But in humid climates and especially under forest cover where the soils. It is probable that the physical properties of other soils are affected also effect of air-drying may be profound. This effect then is determined by the inorganic composition of the soil, when considering the results of the mechanical analyses, and by both the inorganic and organic composition, when considering the consistency values.

PURI and KEEN (1925) already mentioned the fact that dehydration of the

soil by drying, may result in a vigorous cementing action of the clay colloids on soil aggregates. The dispersion of such aggregates requires a rehydration of the clay particles and this rehydration may be very slow, especially when the clay content is large. This rehydration can then be accelerated by boiling and shaking.

In our case, however, the dehydration is, at least partly, irreversible. The clay fraction contains mainly amorphous substances as appeared from X-ray analysis. The mineral counts, represented in table 1, make it probable that these substances will very probably be gel-like silicic acid and volcanio glass. Silicic acid gels have the property of shrinking enormously on drying. During the drying process the gel structure alters appreciably with the result that, on rewetting, the original volume is not reached. Moreover, air has entered the gel structure and this air can hardly be removed on rewetting, so that the dispersion is incomplete (for details of the behaviour of silicic acid gels, see : VAN BEMMELEN (1910) and HERMANS (1949)). A second effect exerted by silicic acid gel in a dry state, is its vigorous cementation action; in this way other soil particles stick together to form larger aggregates. From these considerations it can be concluded that if soils such as those considered here, containing gellike silicic acids are dried at the air, the mechanical composition shifts to the coarse range (as shown in table 3), in the first place because of sticking action of silicic acid and secondly because of the impossibility to redisperse the silicic acid.

The above mentioned mechanism also explains why the consistency values have been lowered by drying the soil and why the plastic number changes from that of a plastic clay soil (> 15) to a coarse to medium sand (< 10).

The change in the consistency values may also be due to a change in the nature of the organic matter. There are several indications that sometimes organic matter cannot be wetted after drying, partly because of the absorption of air (PUCHNER, 1896; EHRENBERG and SCHULTZE, 1914; TSCHAPEK, 1934; LUIKOV, 1935; MEYERS, 1937), partly because of a polymerisation of organic molecules under the influence of Al-, Fe- or Si-compounds (TYULIN, 1940; HUDIG and REDLICH, 1940; HUDIG and SIEWERTSZ VAN REESEMA, 1940), so that a hydrophobic humate is formed. Both phenomena prevent the water from penetrating into the pores of the aggregates and this leads to a lowering of the consistency values.

CONCLUSION

On the basis of the considerations, outlined above, it is advisable not to dry soil samples prior to analysing in the laboratory. The samples have to be analysed in a wet state and every measure must be taken to prevent drying out in the course of the analysis. This is difficult, especially for routine laboratories, but the changes caused by air-drying may be so profound that one has to accept these difficulties, especially for tropical soils, as the author believes to have shown.

ACKNOWLEDGEMENT

The author is indebted to Mr M. SOEPARDI, analytical chemist at the Institute of Soil Research, Bogor, Java, for his help in carrying out the physical analyses.

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rtter %	org, ma content	48.8		35.0	20.3	7.2	4.2
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ncy val weight	point · point ·	344 - 183	I	242 133	 217 	206 - 72 -	256 75
consiste en dry	plastic number	46 18	J	14 19	 04 ∞	5 24 1 0	မ္မာရ
erberg &	plastic limit	227 128	1	181 95	 170 74 -	132	166 56
Atte	biupil timil	273 - 146	I	228 - 104	 210 82 82	186 - 61 -	1 202 65
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	> 50 µ	23.2 21.5 55.0		23.9 19.5 59.2 73.6	26.5 19.3 87.5 90.4	26.1 26.5 86.2 77.3	65.1 64.7 78.2
	Treatment	moist moist, oxidized air-dried	oxidized	moist moist, oxidized air-dried, air-dried, oxidized	moist moist, oxidized air-dried, oxidized	moist moist oxidized air-dried, air-dried, oxidized	moist moist, oxidized air-dried, oxidized
	Profile	I 1-3 cm		3–13 cm	13—41 cm	41–76 cm	76–100 cm

1.6	43.5	21.7	8.1	67 67	27.4
31 28	206 - - -	156 - 64	178 	74 32 1	144 - 60 -
10 1	41 67	3514	51 - -	24 - 11 - 11	29 29
1 48 1	305 	231 - 128 -	265	114 - 52	207 115 -
1 0 1	17	1011	10	31 6	1 10 1
35155	121	168	177 	67 	148 - 79
14 - 14 -	273 - 138 -	210	86	89 I 8 3 I 88	186
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9.1 2.2.7 4.9 not deter	39.9 15.3 12.0 21.0	26.2 16.4 10.9 not det	27.6 25.2 1.7 2.7	12.0 17.2 1.9 5.1	28.0 13.5 10.8 9.4
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moist moist air-dried air-dried, oxidized	moist moist, oxidized air-dried air-dried, oxidized	moist moist, oxidized air-dried, air-dried, oxidized	moist moist, oxidized air-dried air-dried, oxidized	moist moist, oxidized air-dried air-dried, oxidized	moist moist, oxidized air-dried air-dried, oxidized
> 100 cm	II 1- 5 cm	5-25 cm	25100 cm	> 110 cm	III 1.5–16 cm

			Mechanical	compositi minera	on in % o I soil	of oven-dry	-	Atte	therg c	sonsister en dry	ncy val weight	ues	ity olding	atter t %
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oxidized	ŗġ	63.4	6.3	9.I	6.3	6.7	8.0	1	1	1	1	1	<u>,</u> 1	
moist moist air-dried	dized	16.6 16.6 67.9	7.6 7.1 1.9	16.6 15.3 2.4	15.4 21.5 4.9	19.9 30.6 18.4	23.9 8.8 4.8	107 55	8 1 8 1 8	22	$\frac{131}{72}$	23 29 23	78 32	4.1
air-dried, oxidize		70.6	2.6	4.2	4.0	6.4	12.2	I	1	I	-1	1	1	
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